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# REFERENCES

1. Bohlmann, F., Brindöpke, G. and Rastogi, R. C. (1978) *Phytochemistry* 17, 465.
2. Bohlmann, F., Jakupovic, J., Gupta, R. K., King, R. M. and Robinson, H. (1981) *Phytochemistry* 20, 473 (there the change of the absolute configuration is proposed).
3. Bohlmann, F. and Zdero, C. (1982) *Phytochemistry* 21, 2263.
4. Bohlmann, F., Zdero, C., King, R. M. and Robinson, H. (1982) *Phytochemistry* 21, 1045.
5. Bohlmann, F. and Czerson, H. (1978) *Phytochemistry* 17, 1190.
6. Betkouski, M., Mabry, T. J., Taylor, I. F. and Watson, W. H. (1975) *Rev. Latinoam. Quim.* 6, 191.
7. Mabry, T. J., Abdel-Baset, Z., Padolina, W. G. and Jones, S. B. (1975) *Biochem. Syst. Ecol.* 2, 185.

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## A HYDROXYGERMACRENE AND OTHER CONSTITUENTS FROM *PSEUDOBICKELLIA BRASILIENSIS*\*

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**Key Word Index**—*Pseudobrickellia brasiliensis*; Compositae; sesquiterpene; 4 $\beta$ -hydroxygermacra-1(10),5-diene; triterpenes; 11 $\alpha$ -hydroxy- $\alpha$ -amyrin.

**Abstract**—*Pseudobrickellia brasiliensis* afforded, in addition to known compounds, a new germacradiene derivative and a hydroxy- $\alpha$ -amyrin.

The small genus *Pseudobrickellia* is placed in the subtribe Alomiinae (tribe Eupatorieae) [1]. So far nothing is known on the chemistry of this genus. The aerial parts of *P. brasiliensis* (Spreng.) K. et R. afforded lupeol and its  $\Delta^{12}$  isomer,  $\beta$ -amyrin acetate, spathulenol, cadinene, cadinol, oplopanone (1) [2] and a further sesquiterpene alcohol. The structure of the latter followed from the  $^1\text{H}$  NMR spectrum (Table 1) which was very close to that of 2 [3]. However, the chemical shifts of H-5 and H-6 had changed, the double doublet of H-6 now being at higher field. Nuclear Overhauser experiments showed by irradiation of the signal of H-15 a clear effect on the signals of H-3 $\beta$  and H-5. Inspection of models showed that obviously the preferred conformations were the same for both 2 and 3, a chair-chair conformation with the 10-methyl and the 4-methyl (in 2) or 4-hydroxyl group (in 3) quasi-axial above the plane. This clearly followed from the couplings observed. Accordingly, the new sesquiterpene alcohol is the 4-epimer of 2 with a quasi-axial hydroxyl at C-4. Furthermore two isomeric triterpene diols were present

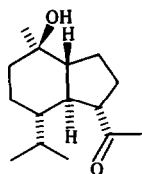
Table 1.  $^1\text{H}$  NMR spectral data of compounds 2 and 3 (400 MHz,  $\text{CDCl}_3$ , TMS as internal standard)

	3	2	2 ( $\text{C}_6\text{D}_6$ )
H-1	4.95 d br	4.95 d br	4.97 d br
H-2 $\alpha$	1.95 d br	1.96 d br	1.96 m
H-2 $\beta$	2.50 dddd	2.51 dddd	2.67 dddd
H-3 $\alpha$	1.54 m*	1.52 m	1.35 m
H-3 $\beta$	1.64 ddd	1.65 ddd	1.50 ddd
H-5	5.25 d	5.17 d	5.06 d
H-6	5.17 dd	5.25 dd	5.30 dd
H-7	2.02 m	2.02 m	1.96 m
H-8	1.39 m	1.41 m	1.35 m
H-9	2.25 m	2.26 m	2.21 m
H-11	1.39 m	1.41 m	1.35 m
H-12	0.82 d	0.84 d	0.95 d
H-13	0.78 d	0.80 d	0.91 d
H-14	1.54 s br	1.55 s br	1.59 dd
H-15	1.19 s	1.21 s	1.12 s

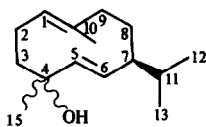
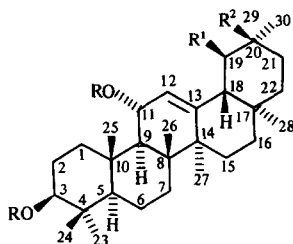
\* $\text{CDCl}_3$ - $\text{C}_6\text{D}_6$ , 2:1, H-3 $\alpha$  1.40 ddd.

$J$  (Hz): 1, 2 $\alpha$  ~ 2.5; 1, 2 $\beta$  = 11.5; 2 $\alpha$ , 2 $\beta$  = 14; 2 $\alpha$ , 3 $\alpha$  ~ 3; 2 $\alpha$ , 3 $\beta$  = 3.5; 2 $\beta$ , 3 $\alpha$  = 11; 2 $\beta$ , 3 $\beta$  = 3.5; 3 $\alpha$ , 3 $\beta$  = 14; 5, 6 = 16; 6, 7 = 9.3; 11, 12 = 11, 13 = 7.

\*Part 461 in the series "Naturally Occurring Terpene Derivatives". For part 460 see Greger, H., Zdero, C. and Bohlmann, F. (1983) *Phytochemistry* 21, 2085.



1

2  $\alpha$ OH  
3  $\beta$ OH

- 4  $R = R^1 = H, R^2 = Me$   
 5  $R = R^2 = H, R^1 = Me$   
 6  $R = Ac, R^1 = H, R^2 = Me$   
 7  $R = Ac, R^2 = H, R^1 = Me$

Table 2.  $^1H$  NMR spectral data of compounds 6 and 7 (400 MHz,  $C_6D_6$ , TMS as internal standard)

	6	7
H-3	4.78 <i>dd</i>	4.75 <i>dd</i>
H-9	2.02 <i>d</i>	1.93 <i>d</i>
H-11	5.77 <i>dd</i>	5.73 <i>dd</i>
H-12	5.60 <i>d</i>	5.45 <i>d</i>
Me	1.36 <i>s</i>	1.21 <i>s</i>
	1.03 <i>s</i> (6H)	1.00 <i>s</i>
	0.97 <i>s</i>	0.98 <i>s</i>
	0.96 <i>s</i>	0.93 <i>s</i>
	0.95 <i>s</i> (6H)	0.91 <i>s</i>
	0.87 <i>s</i>	0.88 <i>s</i>
		0.98 <i>d</i>
		0.93 <i>d</i>
OAc	1.80 <i>s</i>	1.79 <i>s</i>
	1.74 <i>s</i>	1.78 <i>s</i>

$J$  (Hz): 2, 3 = 11; 2', 3 = 5; 9, 11 = 8.5; 11, 12 = 3.5; compound 7: 19, 29 = 20, 30 = 6.

which, however, could only be separated as their diacetates by HPLC (reversed phase). One of these diacetates was identical with 11 $\alpha$ -acetoxy- $\beta$ -amyrin acetate (6) [4] while the second one showed two methyl doublets in the  $^1H$  NMR spectrum (Table 2). The other signals were close to those of 6 especially those of H-9, H-11 and H-12. Therefore the second diacetate most probably is 7. The minute amount did not allow chemical transformations to establish the proposed structure. However, the similarity of the  $^1H$  NMR spectra of 6 and 7 and comparison with the spectra of  $\alpha$ - and  $\beta$ -amyrin acetate strongly supported the structure which also was in agreement with the mass spectral fragmentation pattern.

## EXPERIMENTAL

The air dried aerial parts (320 g) (voucher T. S. F. 892) were extracted with  $Et_2O$ -petrol, 1:2, and the resulting extract was separated by CC (silica gel) and further by repeated TLC (silica gel). The petrol fraction gave 5 mg  $\gamma$ -cadinene, that one with  $Et_2O$ -petrol (1:10) 20 mg  $\beta$ -amyrin acetate, that one with  $Et_2O$ -petrol (1:3), 150 mg lupeol, 150 mg of its  $\Delta^{12}$  isomer, 10 mg spathulenol, 10 mg  $\alpha$ -cadinol, 20 mg 1 and 20 mg 3 while that one with  $Et_2O$  after acetylation ( $Ac_2O$ , 1 hr, 70°) afforded a mixture of 3 mg 6 and 3 mg 7 ( $Et_2O$ -petrol, 1:3) which was separated by HPLC (reversed phase,  $MeOH-H_2O$ , 50:1).

Compound 6. Colourless crystals, mp. 214° (lit. 216° [4]);  $^1H$  NMR: see Table 2.

Compound 7. Colourless gum, IR  $\nu_{max}^{CCl_4}$   $cm^{-1}$ : 1730, 1250

(OAc); MS  $m/z$  (rel. int.): 526.402  $[M]^+$  (3) ( $C_{34}H_{54}O_4$ ), 466  $[M - HOAc]^+$  (100), 451  $[466 - Me]^+$  (12), 406  $[466 - HOAc]^+$  (6), 391  $[406 - Me]^+$  (12), 276  $[C_{18}H_{28}O_2, RDA]^+$  (14), 234  $[276 - ketene]^+$  (20).

$$[\alpha]_{24}^{20} = \frac{589}{-26} \frac{578}{-27} \frac{546}{-32} \frac{436}{-58} \text{ nm} \text{ (c 0.2, } CHCl_3\text{)}.$$

4 $\beta$ -Hydroxygermacra-1(10),5-diene (3). Colourless oil, bp<sub>0.1 Torr</sub> 110°; IR  $\nu_{max}^{CCl_4}$   $cm^{-1}$ : 3603 (OH), 3060, 1630, 980 ( $CH=CH$ ); MS  $m/z$  (rel. int.): 222.198  $[M]^+$  (2) ( $C_{15}H_{26}O$ ), 207  $[M - Me]^+$  (5), 204  $[M - H_2O]^+$  (3), 189  $[207 - H_2O]^+$  (6), 161  $[204 - C_3H_7]^+$  (13), 81  $[C_6H_9]^+$  (100).

$$[\alpha]_{24}^{20} = \frac{589}{-109} \frac{578}{-116} \frac{546}{-133} \frac{436}{-235} \text{ nm} \text{ (c 0.62, } CHCl_3\text{)}.$$

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## REFERENCES

- King, R. M. and Robinson, H. (1980) *Phytologia* 46, 446.
- Takeda, K., Minato, H. and Ishikawa, M. (1966) *Tetrahedron Suppl.* 7, 219.
- Bohlmann, F., Knoll, K.-H., Zdero, C., Mahanta, P. K., Grenz, M., Suwita, A., Ehlers, D., Le Van, N., Abraham, W.-R. and Natsu, A. A. (1977) *Phytochemistry* 16, 965.
- Taylor, D. A. H. (1967) *J. Chem. Soc. (C)* 490.